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# Nigroxanthin (3',4'-Didehydro- $\beta$ , $\gamma$ -carotene-3,6'-diol), a New Carotenoid Isolated from Paprika (*Capsicum annuum* var. *longum nigrum*)

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**Abstract.** From red paprika (*Capsicum annuum* var. *longum nigrum*) nigroxanthin (**1**) was isolated as a minor carotenoid and, based on its spectral data, identified as (all-*E*)-3',4'-didehydro- $\beta$ , $\gamma$ -carotene-3,6'-diol.

## Introduction

The different varieties of paprika (*Capsicum annuum*) have been investigated for a long time. It has been established that capsanthin ((3*R*,3'*S*,5'*R*)-3,3'-dihydroxy- $\beta$ , $\kappa$ -caroten-6'-one) and capsorubin ((3*S*,5*R*,3'*S*,5'*R*)-3,3'-dihydroxy- $\kappa$ , $\kappa$ -carotene-6,6'-dione), both of which contain the five membered ring  $\kappa$ -end group, are the most abundant carotenoids in these vegetables [1]. Furthermore, many other carotenoids with interesting structures, especially those with the oxabicyclo- $\beta$ -end group, have been isolated [2][3]. Recently, the carotenoid composition of the different paprika varieties at different stages of ripening was under investigation [4][5]. This was done in view of the elucidation of

the biosynthesis of the  $\kappa$ -end group, which still has not been completely established.

During the investigation of the black variety (*Capsicum annuum* var. *longum nigrum*) 58 peaks were observed by HPLC, and 34 carotenoids were completely or tentatively identified [5]. In this paper, as a continuation of these studies, the isolation of a hitherto unknown carotenoid, the assignment of its constitution and its configuration at the double bonds is reported.

## Results and Discussion

During the isolation of cycloviolaxanthin [3], several unknown carotenoids were observed by column chromatography. The compound which was absorbed between cucurbitaxanthin A and B (zone 7 in [3], and peak 30 in Fig. 1 in [5]) on the CaCO<sub>3</sub> column (*Biogal*, Hungary) was further investigated. From this zone, a new carotenoid, for which the name '*Nigroxanthin*' (**1**) is proposed, was isolated and crystallized from benzene/hexane (m.p. 125–127°).

The UV/VIS spectrum ( $\lambda_{\text{max}}$ , benzene: 487, 457, and 434 nm, no *cis* peak) shows that the compound contains an (all-*E*)-decaene chromophore. In accordance with that, no reaction took place with LiAlH<sub>4</sub> or HCl/AcOH indicating that no carbonyl or 5,6-epoxy groups are present. The EI-MS

shows the signal for the molecular ion at  $m/z$  566 (100,  $M^+$ ) which corresponds to C<sub>40</sub>H<sub>54</sub>O<sub>2</sub>. Further characteristic signals can be observed at 548 ( $[M - H_2O]^+$ ), 530 ( $[M - 2H_2O]^+$ ), 474 ( $[M - 92]^+$ ), 119, and 105. Characteristic signals for allenes, acetylenes, carbonyl, carboxyl, and epoxy groups were absent in the IR spectrum. Acetylation gave a crystalline compound with a molecular ion at  $m/z$  608 (100) in the MS corresponding to a monoacetate. The reaction with (CH<sub>3</sub>)<sub>3</sub>SiCl/((CH<sub>3</sub>)<sub>3</sub>Si)<sub>2</sub> resulted in a mono-trimethylsilyl ether (MS: 638 ( $M^+$ )). These derivatives give an indication that nigroxanthin (**1**) contains one *prim.* or *sec.* and one *tert.* OH group.

For the NMR investigations, the compound was once again recrystallized. The HPLC analysis of this purified compound showed a purity of > 98%. The analysis of the <sup>1</sup>H- and <sup>13</sup>C-NMR spectra of nigroxanthin (**1**) was restricted to the end group signals, as the application of modern techniques is hampered by the small sample concentration. However, larger quantities of deoxylutein II (**2**) and deoxylutein III (**3**) were available to perform the experiments necessary for complete structural elucidation. Taking advantage for their structural relationship to **1**, their data serve for further signal identifications in the spectra of nigroxanthin.

NMR Investigations of both carotenoids **2** and **3** have been already published [6], but as we obtained more detailed spectral informations and found different line assignments in some cases, all NMR spectroscopic data for **1–3** are shown in the *Table*. Not to overload the table, only relevant <sup>2</sup>*J*(H,H) values are listed. The <sup>1</sup>H and <sup>13</sup>C resonances of the well-known  $\beta$ -end groups can clearly be assigned on the basis of their chemical-shift values and coupling interactions, and are in agreement with data from [7][8].

The <sup>1</sup>H-NMR resonances for the geminal protons H<sub>2</sub>C(2') in the  $\gamma$ -end group are identified due to their chemical shifts of 2.30 and 2.22 ppm. In the H,H-COSY spectrum cross-peaks between H<sub>2</sub>C(2') and H-C(3') as well as between H-C(3') and H-C(4') are visible, so that the signals for both olefinic protons can be assigned. Two slightly broadened singlets, with  $\delta$  values (5.03 and 5.00 ppm) typical for exocyclic olefinic CH<sub>2</sub> protons, correspond to the nuclei H<sub>2</sub>C(18'), which may be arbitrarily named H<sub>a</sub> and H<sub>b</sub>. Their very small coupling interaction, causing signal broadening, is detectable in the COSY spectrum. For deoxylutein II (**2**), it can be verified by 1D NOE difference experiments that the signal appearing at lower field corresponds to H<sub>a</sub> (NOE H<sub>a</sub>  $\leftrightarrow$  H-C(4')), and the reso-

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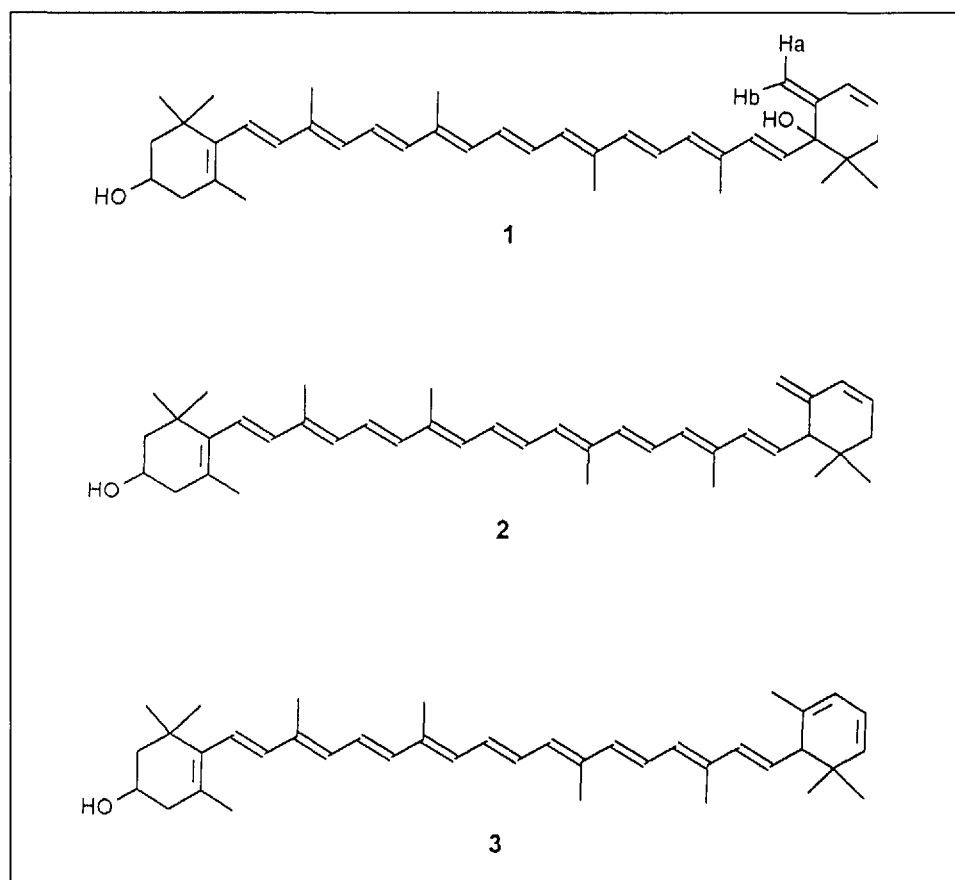
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Table.  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR Data for 1, 2, and 3

C	$^1\text{H}$			$^{13}\text{C}$		
	1	2	3	1	2	3
1	–	–	–	37.13	37.12	37.12
2	1.77 eq, 1.48 ax	1.77 eq, 1.48 ax	1.77 eq, 1.48 ax	48.46	48.44	48.44
3	4.00, OH: 1.56	4.00, OH: 1.48	4.00, OH: 1.40	65.10	65.09	65.09
4	2.39 eq, 2.05 ax	2.39 eq, 2.02 ax	2.39 eq, 2.04 ax	42.58	42.56	42.56
5	–	–	–	126.16	126.16	126.15
6	–	–	–	137.77	137.77	137.77
7	6.10 (AB)	6.10 (AB)	6.11 (AB)	<sup>e)</sup>	125.55	125.55
8	6.15 (AB)	6.15 (AB)	6.15 (AB)	138.51	138.51	138.51
9	–	–	–	135.57 <sup>a)</sup>	135.47 <sup>a)</sup>	135.63
10	6.19 (10.4)	6.17 (12.2)	6.16 (12.0)	131.32	131.32	131.32
11	6.65 (10.4, 14.8)	6.64 (12.2, 14.9)	6.63 (12.0, 14.9)	124.90 <sup>b)</sup>	124.87	124.86
12	6.36 (14.8)	6.36 (14.9)	6.36 (14.9)	137.58	137.59	137.59
13	–	–	–	136.44	136.39	136.37
14	6.25 <sup>a)</sup>	6.25	6.26	<sup>e)</sup>	132.62 <sup>b)</sup>	132.62
15	6.64	6.61	6.63	<sup>e)</sup>	130.10 <sup>c)</sup>	130.11 <sup>a)</sup>
16	1.07	1.07	1.07	28.73	28.73	28.72
17	1.07	1.07	1.07	30.26	30.26	30.26
18	1.71	1.74	1.73	21.82	21.61	21.62
19	1.97	1.97	1.97	12.82	12.81	12.81
20	1.97	1.97	1.97	12.82	12.81	12.81
1'	–	–	–	49.81	33.45	34.81
2'	2.30, 2.22 (AB)	1.94, 2.01 (AB)	5.32 (9.4)	29.70	38.52	134.53
3'	5.67 (10.1, 3.7)	5.71 (10.1, 4.7)	5.75 (9.4, 5.1)	<sup>e)</sup>	127.81	122.10
4'	6.26	6.18	5.60 (5.1)	<sup>e)</sup>	128.30	117.55
5'	–	–	–	149.32	145.68	137.22
6'	–	2.65 (9.2)	2.20 (10.0)	73.67	55.49	55.72
7'	<sup>e)</sup>	5.64 (15.5, 9.2)	5.57 (15.1, 10.0)	<sup>e)</sup>	129.24	128.00
8'	<sup>e)</sup>	6.16 (15.5)	6.12 (15.5)	<sup>e)</sup>	136.59	135.88
9'	–	–	–	135.66 <sup>a)</sup>	135.64 <sup>a)</sup>	135.73
10'	6.13 (10.5)	6.16 (11.8)	6.15 (11.7)	<sup>e)</sup>	130.66	130.46
11'	6.64 (10.5, 14.8)	6.61 (11.8, 15.0)	6.60 (11.7, 15.0)	124.94 <sup>b)</sup>	124.98	125.01
12'	6.36 (14.8)	6.34 (15.0)	6.33 (15.0)	137.49	137.28	137.14
13'	–	–	–	136.47 <sup>c)</sup>	136.51	136.53
14'	6.26 <sup>a)</sup>	6.24	6.25	<sup>e)</sup>	132.40 <sup>b)</sup>	132.35
15'	6.64	6.61	6.63	<sup>e)</sup>	129.95 <sup>c)</sup>	129.92 <sup>a)</sup>
16'	1.13 <sup>b)</sup>	0.88 <sup>a)</sup>	0.93	<sup>e)</sup>	25.33 <sup>d)</sup>	25.82
17'	1.22 <sup>b)</sup>	0.90 <sup>a)</sup>	1.00	<sup>e)</sup>	28.42 <sup>d)</sup>	26.66
18'	5.03 H <sub>a</sub> , 5.00 H <sub>b</sub>	4.88 H <sub>a</sub> , 4.81 H <sub>b</sub>	1.72	105.09	112.97	22.17
19'	1.97	1.92	1.88	13.22	13.13	13.12
20'	1.97	1.96	1.96	12.76	12.75	12.75

Chemical shift values:  $\delta$  [ppm]  $J(\text{H,H})$  coupling constant values: ( $J$  [Hz]).<sup>a)</sup>–<sup>d)</sup> Assignment may be interchanged.<sup>e)</sup> Not assigned.



nance appearing at higher field to  $H_b$  (NOE  $H_b \leftrightarrow H-C(7')$ ). Analogously the singlets at 5.03 and 5.00 ppm in the spectrum of nigroxanthin (**1**) were assigned to  $H_a$  and  $H_b$ .

The resonances for  $H_3C(16')$  and  $H_3C(17')$  can be found in the expected spectral region, but, as in the case of **2**, their assignments may be exchanged. The signal for  $HO-C(6')$  was not identified.

The  $^{13}C$  resonances of  $C(1')$ ,  $C(2')$ ,  $C(5')$ , and  $C(6')$  can be identified directly from their characteristic  $\delta$  values and informations out of the DEPT-135 experiment, while line assignment for  $C(3')$ ,  $C(4')$ ,  $C(16')$ , and  $C(17')$  is not feasible without any doubt.  $C(18')$  gives a resonance at 105.09 ppm, typically for an exocyclic  $CH_2-sp^2-C$ -nucleus. It must be mentioned that its relative signal intensity is unexpectedly weak; reasons for this observation are unknown.

Applications of H,H-COSY, C,H-shift correlation and COLOC experiments to **2** and **3** allow extensive line assignments for  $^1H$  and  $^{13}C$  resonances. Based on comparison with these data sets, most of the  $^1H$  and a number of the  $^{13}C$  signals of the nigroxanthin olefinic chain can be assigned. Inspection of the  $^nJ(H,H)$  values confirms the supposed (all-*E*)-configuration of **1**.

Based on the spectroscopic data, especially the NMR investigations, nigroxan-

thin (**1**) was identified as (all-*E*)-3',4'-dihydro- $\beta,\gamma$ -carotene-3,6'-diol. The configuration at  $C(3)$  and  $C(6')$  remains at the moment unknown. The total synthesis of optically active compounds with the constitution of **1** is in progress.

### Experimental Part

**General.** HPLC: Gynkotek pump model 300 B with Gynkotek gradient former, detector: Waters-991, photo diode array. Column: 250 x 4.6 mm i.d., *chromsyl C<sub>18</sub>*, 6  $\mu$ m, endcapped. Mobile phase: eluent A 12%  $H_2O$  in MeOH, eluent B: MeOH, eluent C: acetone/MeOH 1:1. Gradient program: 0–2 min: 100% A; –10 min: to 80% A/20% B; –18 min: to 50% A/50% B; –25 min: to 100% B; –27 min: 100% B, –34 min: to 100% C; –41 min 100% C (linear steps). UV/VIS: Beckman DU-65. CD: Jobin-Yvon Dichrograph-6 in MeOH at r.t. NMR: Varian Unity 300 ( $^1H$ : 300 MHz,  $^{13}C$ : 75.43 MHz), Bruker AM 400 ( $^1H$ : 400.14 MHz,  $^{13}C$  100.61 MHz), Bruker AC 300 ( $^1H$ : 300.13 MHz,  $^{13}C$  75.47 MHz) 0.2 mg of **1**, 11 mg of **2**, 6 mg of **3** in 0.5 ml of  $CDCl_3$  at 20°. Chemical shifts ( $\delta$ ) in ppm (relative to the solvent signal), coupling constants (*J*) in Hz. MS: Jeol JMS-01-SG-2.

**Isolation.** A detailed description of the general isolation procedure has been given in [3]. After desorption of zone 7 [3], nigroxanthin (**1**) was precipitated in benzene/hexane (12 mg, m.p. 99°, purity 86%). After recrystallization (benzene/hexane) the product was submitted to CC: 3 columns 6 x 30 cm,  $CaCO_3$  (Biogal, Hungary), 3–4% acetone in hexane. Picture after development:

30 mm yellow: unknown; 60 mm intermediate zone; 30 mm yellow (nigroxanthin (**1**)); 8 mm rose-colored: unknown, 10 mm intermediate zone, 5 mm yellow: unknown, 20 mm intermediate zone, 20 mm: pale yellow. After the development the column was extruded and cut into pieces. After desorption **1** was crystallized (benzene/hexane) to give 8 mg of red crystals (m.p. 125–127°, purity > 98%. DC:  $R_f$  0.64 (silica  $F_{254}$  (Merck 5554), benzene/AcOEt/MeOH 7:2:1).

**Spectroscopical Data of 1.** IR (KBr): 883s, 3425m. UV/VIS (benzene): 487 (1.157), 457 (1.293), 434 (0.880). UV (hexane): 472, 441, 421. UV ( $Et_2O$ ): 473, 445. UV (MeOH): 472, 444. UV (EtOH): 474, 447. MS: 566 (100,  $M^+$ ), 548 (11), 530 (2), 474 (9), 119 (28), 105 (22).  $^1H$ - and  $^{13}C$ -NMR: cf. the Table.

**Nigroxanthin monoacetate.** Acetylation according to [9]. Crystallization: benzene/MeOH; m.p. 82–84°. DC:  $R_f$  0.87 (silica gel  $F_{254}$  (Merck 5554), benzene/AcOEt/MeOH 7:2:1). IR (KBr): 3420m. UV/VIS (benzene): 487, 457, 433. EI-MS: 608 (100,  $M^+$ ,  $C_{42}H_{56}O_3$ ), 548 (5), 516 (15), 502 (3), 456 (4).

**Nigroxanthin Trimethylsilyl Ether.** Silylation according to [10] DC:  $R_f$  0.85 (silica gel  $F_{254}$ , (Merck 5554), benzene/AcOEt 7:3). IR (KBr): 895s, 1715s, 3470m. UV/VIS (benzene): 487, 457, 433. EI-MS: 638 (35,  $M^+$ ,  $C_{43}H_{62}O_2Si$ ), 566 (100), 548 (4), 546 (15), 368 (12).

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